# 5,6-Dimethoxyindole 

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#### Abstract

C}_{10} \mathrm{H}_{11} \mathrm{NO}_{2}, M_{r}=177 \cdot 21\), orthorhombic, $P 2,2,2, \quad a=5.805(1), \quad b=11 \cdot 195$ (2), $\quad c=$ 13.533 (2) $\AA, \quad V=879.5$ (2) $\AA^{3}, \quad Z=4, \quad D_{x}=$ $1.34 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \mathrm{Cu} K \alpha$ radiation, $\lambda=1.5418 \AA, \quad \mu=$ $7.7 \mathrm{~cm}^{-1}, F(000)=376, T=293 \mathrm{~K}$, final $R=0.036$ for 693 observed reflections. The molecule is nearly planar with the exception of $\mathrm{C}(11)$ which is out of the plane by $-0.257(4) \AA$. The torsion angles $\mathrm{C}(11)-$ $\mathrm{O}(2)-\mathrm{C}(4)$ and $\mathrm{C}(10)-\mathrm{O}(1)-\mathrm{C}(6)-\mathrm{C}(7)$ are -6.0 (5) and $-1.5(5)^{\circ}$, respectively.


Experimental. The title compound (Fig. 1) was purchased from the Aldrich Chemical Company. Crystals grown from acetonitrile solution. Data collected on an Enraf-Nonius CAD-4 diffractometer, graphite monochromator. The crystal had dimensions $0.2 \times 0.3 \times 0.3 \mathrm{~mm}$. Cell parameters measured on the diffractometer using 25 reflections in the $2 \theta$ range $20-40^{\circ}$. Range of indices $0<h<6,0<k<12$, $0<l<15\left(\theta<60^{\circ}\right)$. Three standards 101, 024, 131, measured after every 200 reflections, showed a variation of $0.2 \%$. No absorption corrections. Lorentz and polarization corrections. 785 unique reflections measured; 693 observed reflections with $I>3 \cdot 0 \sigma(I)$. Direct methods (MULTAN11/82; Main, Fiske, Hull,


Fig. 1. Numbering of atoms and conformation of the molecule.

Lessinger, Germain, Declercq \& Woolfson, 1982) used for structure determination. H atoms located by difference Fourier synthesis. Anisotropic full-matrix least-squares refinement (on $F$ ) for non- H atoms,

Table 1. Final fractional coordinates and equivalent isotropic temperature factors for non- H atoms with e.s.d.'s in parentheses

$$
B_{\mathrm{eq}}=\frac{4}{3} \sum_{i} \sum_{j} \boldsymbol{B}_{l j} \mathbf{a}_{i} \cdot \mathbf{a}_{j}
$$

|  | $\boldsymbol{x}$ | $y$ | $z$ | $B_{\text {eq }}\left(\AA^{2}\right)$ |
| :--- | ---: | ---: | ---: | ---: |
|  |  |  |  |  |
| $\mathrm{O}(1)$ | $0.3447(5)$ | $-0.0990(2)$ | $-0.0039(2)$ | $4.00(5)$ |
| $\mathrm{O}(2)$ | $0.0080(5)$ | $-0.0059(2)$ | $-0.1005(2)$ | $4.14(5)$ |
| $\mathrm{N}(1)$ | $0.1533(6)$ | $0.1664(3)$ | $0.2746(2)$ | $4.13(7)$ |
| $\mathrm{C}(2)$ | $-0.0167(8)$ | $0.2515(4)$ | $0.2815(3)$ | $4.49(9)$ |
| $\mathrm{C}(3)$ | $-0.1350(7)$ | $0.2588(3)$ | $0.1945(3)$ | $4.43(8)$ |
| $\mathrm{C}(4)$ | $-0.0852(6)$ | $0.1341(3)$ | $0.0327(3)$ | $3.61(8)$ |
| $\mathrm{C}(5)$ | $0.0439(6)$ | $0.0434(3)$ | $-0.0077(2)$ | $3.20(7)$ |
| $\mathrm{C}(6)$ | $0.2260(6)$ | $-0.0097(3)$ | $0.0457(2)$ | $3.23(7)$ |
| $\mathrm{C}(7)$ | $0.2778(7)$ | $0.0248(3)$ | $0.1413(2)$ | $3.43(7)$ |
| $\mathrm{C}(8)$ | $0.1410(7)$ | $0.1161(3)$ | $0.1811(2)$ | $3.37(7)$ |
| $\mathrm{C}(9)$ | $-0.0356(7)$ | $0.1714(3)$ | $0.1295(2)$ | $3.48(7)$ |
| $\mathrm{C}(10)$ | $0.5318(7)$ | $-0.1534(3)$ | $0.0488(3)$ | $4.65(9)$ |
| $\mathrm{C}(11)$ | $-0.1929(7)$ | $0.0354(4)$ | $-0.1519(3)$ | $4.52(9)$ |



Fig. 2. Molecular packing diagram.

Table 2. Bond distances $(\AA)$ and bond angles $\left({ }^{\circ}\right)$ with e.s.d.'s in parentheses

| $\mathrm{O}(1)$ | C (6) | 1.387 (4) | C(3) | C(9) | 1.436 (5) |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O}(1)$ | $\mathrm{C}(10)$ | 1.434 (5) | C(4) | C(5) | 1.375 |  |
| $\mathrm{O}(2)$ | C(5) | 1.387 (4) | C(4) | C(9) | 1.404 |  |
| $\mathrm{O}(2)$ | $\mathrm{C}(11)$ | 1.433 (5) | C(5) | C(6) | 1.411 |  |
| $\mathrm{N}(1)$ | C(2) | 1.373 (5) | C(6) | C(7) | 1.383 |  |
| $\mathrm{N}(1)$ | C(8) | 1.387 (4) | C(7) | C(8) | 1.402 |  |
| C(2) | $\mathrm{C}(3)$ | 1.365 (5) | C(8) | C(9) | 1.385 |  |
| C(6) | $\mathrm{O}(1)$ | $\mathrm{C}(10) 116.2$ (3) | O(1) | C(6) | C(7) | 123.1 (3) |
| C(5) | $\mathrm{O}(2)$ | C(11) 115.7(3) | C(5) | C(6) | C(7) | 121.7 (3) |
| C(2) | $\mathrm{N}(1)$ | C(8) 107.9 (3) | C(6) | $\mathrm{C}(7)$ | C (8) | 116.1 (3) |
| N(1) | $\mathrm{C}(2)$ | C(3) 110.1(3) | N(1) | C (8) | C(7) | 128.1 (3) |
| $\mathrm{C}(2)$ | $\mathrm{C}(3)$ | C(9) 106.6 (3) | N(1) | $\mathrm{C}(8)$ | C(9) | 108.5 (3) |
| C(5) | $\mathrm{C}(4)$ | C(9) 118.6 (3) | C(7) | C(8) | C(9) | 123.4 (3) |
| $\mathrm{O}(2)$ | C(5) | C(4) 124.9 (3) | C(3) | C(9) | C(4) | 133.7 (3) |
| $\mathrm{O}(2)$ | C(5) | C(6) 114.1(3) | C(3) | C(9) | C(8) | 107.0 (3) |
| C(4) | C(5) | C(6) 120.9 (3) | C(4) | C(9) | C(8) | 119.3 (3) |
| $\mathrm{O}(1)$ | C(6) | C(5) 115.3 (3) |  |  |  |  |

isotropic for H atoms. $\sum w\left(\left|F_{o}\right|-\left|F_{c}\right|\right)^{2}$ minimized. $R=w R=0.036$, max. $\Delta / \sigma=0 \cdot 12$. Max. peak height in the final difference Fourier map $0.28 \mathrm{e} \AA^{-3}, S$ $=0.51$. Atomic scattering factors from International Tables for X-ray Crystallography (1974). EnrafNonius SDP (Frenz, 1984).

Atomic parameters are given in Table 1,* bond distances and bond angles are presented in Table 2. Atomic numbering is shown in Fig. 1 and packing in Fig. 2.

Related literature. The methoxy group has the same non-planar conformation as in 5,6-dimethoxy-1-indanone (Shoja, 1988).

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[^1]
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